

Letter

Enantioselective palladium-catalyzed allylic substitution with 1-diphenylphosphino-4-dialkylamino ligands

Fabien Robert, Nicolas Gaillard, Denis Sinou *

Laboratoire de Synthèse Asymétrique, associé au CNRS, CPE Lyon, Université Claude Bernard Lyon 1, 43, Boulevard du 11 novembre 1918, 69622 Villeurbanne Cedex, France

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Abstract

Various chiral 1-diphenylphosphino-4-dialkylamino ligands were easily prepared from tartaric acid. Palladium complexes of these ligands gave enantioselectivities up to 75% in the alkylation of 1,3-diphenylprop-2-enyl acetate with dimethyl malonate. © 1999 Elsevier Science B.V. All rights reserved.

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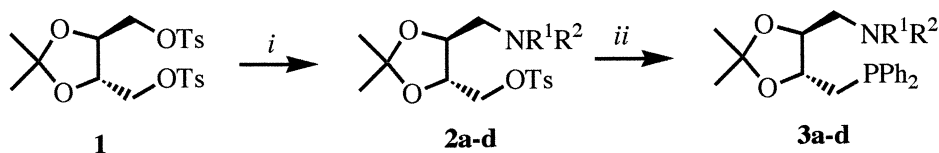
The palladium-catalyzed asymmetric allylic substitution reaction has been shown to be useful in the synthesis of various chiral molecules [1–7]. A large number of chiral ligands have been studied in this allylic substitution. The chiral ligands that produce high enantioselectivities could be divided into three families:

- ligand with a chiral side chain able to direct the attack of the nucleophile to preferentially one atom [8,9]
- ligand with a chiral environment constituting a chiral pocket [10,11]
- ligand exhibiting an electronic desymmetrization such as ‘hetero-chelates’ [12–17].

In the last family, the most studied ligands are *N,P*-chelates such as (phosphinoaryl)-oxazoline ligands, which are generally very efficient in the palladium-catalyzed allylic substitution, giving very high activities as well as enantioselectivities. Surprisingly, only three papers concerning the use of chiral hetero-chelates containing a dialkylamino and a diphenylphosphino group appeared until now in the literature [18–22]. In most of the examples, the chirality was on the substituents of the nitrogen via a binaphthyl unit.

In a programme concerning the synthesis and application of chiral ligands bearing a sp^3 -nitrogen and a diphenylphosphino group, we evaluated the potential utilities of a number of these chiral ligands for enantioselective palladium-catalyzed allylic substitution. We report in this Letter the synthesis of a number of these

* Corresponding author. Tel.: +33-4-72-446263; Fax: +33-4-72-446263; E-mail: sinou@univ-lyon1.fr



i: LiNR¹R² *ii*: LiPPh₂

a: R¹ = R² = C₆H₅; **b**: R¹ = C₆H₅, R² = CH₃; **c**: R¹ = C₆H₄-*p*-OCH₃, R² = CH₃
d: R¹ = C₆H₅, R² = α-naphtyl

Scheme 1.

ligands and their applications in the enantioselective palladium allylic substitution of 1,3-diphenylprop-2-enyl acetate with dimethylmalonate.

The ligands **3a–d** were prepared from the commercial bistosylate **1**. Reaction of **1** at 0°C for 4 h with lithium arylalkylamine or phenyl-arylamine, obtained by treatment of the corresponding amine with lithium diisopropylamine at –78°C for 1 h, gave the aminotosylate **2a–d**. Treatment of **2a–d** with lithium diphenylphosphine at room temperature in THF afforded ligands **3a–d** (see Scheme 1).

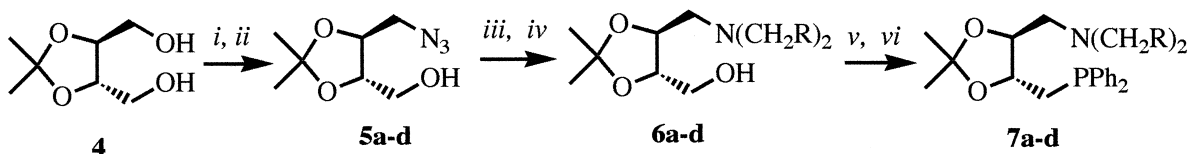
Monotosylation of diol **4** followed by reaction with sodium azide in DMF at reflux gave the azido compounds **5a–d**. Reduction of the azido function with molecular hydrogen in the presence of Pd/C, followed by alkylation of the resulting amino alcohol with various benzylic bromides in the presence of sodium carbonate afforded the dibenzylamino alcohols **6a–d**. Subsequent tosylation of compounds **6a–d** gave the

corresponding aminotosylates, whose treatment with lithium diphenylphosphine at room temperature gave the corresponding aminophosphines **7a–d** (see Scheme 2).

The diaminoligand **8** (see Fig. 1) was also obtained by reaction of the ditosylate **1** with lithium diphenylamide.

We tested the ability of these new ligands in the palladium-catalyzed allylic substitution of racemic (*E*)-1,3-diphenylprop-2-enyl acetate with dimethyl malonate using an in situ catalyst prepared from Pd₂(dba)₃ and the chiral ligand in THF. The results are summarized in Table 1.

The ligands **DIOP** and **8**, having a C₂ symmetry, exhibited no enantioselectivity in this reaction (entries 1–2). When the reaction was performed using NaH as the base, enantioselectivities of 50, 68, 63, and 69% were obtained with ligands **3a**, **3b**, **3c**, and **3d**, respectively (entries 3–6). It seemed that the presence of two different substituents on the nitrogen gave higher enantioselectivities.



i: TsCl, C₅H₅N; *ii*: NaN₃; *iii*: H₂, Pd/C; *iv*: RCH₂Br, Na₂CO₃; *v*: TsCl, C₅H₅N; *vi*: LiPPh₂

a: R = C₆H₅; **b**: R = C₆H₂-2,4,6-*tri*CH₃; **c**: R = α-naphtyl; **d**: R¹ = C₆H₄-*o*-C₆H₅

Scheme 2.

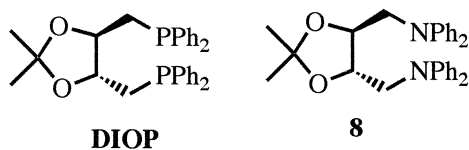


Fig. 1.

In the case of ligand **3d**, the use of BSA as the base in the presence of KOAc in THF gave lower conversion and enantioselectivity (entry 7). However, performing the alkylation in CH_2Cl_2 as the solvent, in the presence of NaH or BSA/KOAc as the base, gave a very active catalyst with enantioselectivity up to 68% (entries 8–9).

The palladium complex generated from $\text{Pd}_2(\text{dba})_3$ and ligand **7** was also an effective catalyst for the asymmetric allylic alkylation of (*E*)-1,3-diphenylprop-2-enyl acetate with the sodium salt of dimethyl malonate, enantiomeric excess up to 75% being obtained (entries 10–

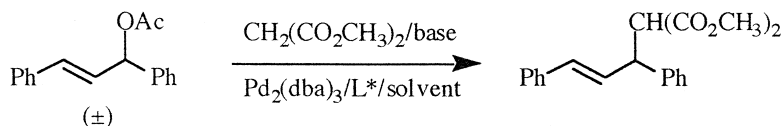
13). However, it seems that the presence of a too bulky group at the ortho position lowered the enantioselectivity of the alkylation reaction (compare entries 10 and 13).

Preliminary experiments concerning the use of ligands **3d** and **7c** associated with $[\text{Rh}(\mu\text{-OMe})(\text{COD})]_2$ in the hydroformylation of styrene at 65°C and 30 atm gave enantioselectivities up to 18 and 15%, respectively, with a *iso/n* ratio of 91/9 and 91/10, respectively [26]; these results are quite promising, since performing the same reaction using Diop as the ligand gave lower regio- and enantioselectivity.

In summary, we have shown that asymmetric palladium-catalyzed allylic alkylation occurred with good enantioselectivities using easily accessible 1-diphenylphosphino-4-dialkylamino ligands derived from tartaric acid. The influence of the introduction of a chirality on the substituents of the nitrogen atom, as well as molecular modelling experiments of the π -allyl intermediate in the way to increase the enantioselectivities are currently under way.

Table 1

Allylic alkylation of (\pm)-(*E*)-1,3-diphenylprop-2-enyl acetate with dimethyl malonate catalyzed by palladium complexes^a



Entry	Ligand	Solvent	Conversion (%) ^b	ee (%) ^b (configuration) ^c
1	DIOP	THF	95	0
2	8	THF	90	0
3	3a	THF	97	50 (<i>S</i>)
4	3b	THF	94	68 (<i>S</i>)
5	3c	THF	96	63 (<i>S</i>)
6	3d	THF	93	69 (<i>S</i>)
7	3d	THF ^d	47	30 (<i>S</i>)
8	3d	CH_2Cl_2	94	67 (<i>S</i>)
9	3d	CH_2Cl_2^d	95	68 (<i>S</i>)
10	7a	THF	94	61 (<i>S</i>)
11	7b	THF	97	75 (<i>S</i>)
12	7c	THF	99	57 (<i>S</i>)
13	7d	THF	88	47 (<i>S</i>)

^a[acetate]/[malonate]/[NaH]/[$\text{Pd}_2(\text{dba})_3$]/[ligand] = 50/150/150/1/2; solvent: THF; 25°C; 24 h.

^bDetermined by HPLC analysis with chiral stationary column Daicel Chiralcel OD-H (hexane/2-propanol = 98/2).

^cDetermination based on the sign of the specific rotation of the alkylated product [23–25].

^d*N,O*-bis[(trimethylsilyl)acetamide] (or BSA) and KOAc were used instead of NaH.

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